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[ISC Computer Sciencefor Class 11](#) Dheeraj Mehrotra ISC Computer Sciencefor Class 11  
[Chemical News and Journal of Industrial Science](#) 1913  
*Practical/Laboratory Manual Chemistry Class - XI* Er. Meera Goyal 2021-05-29 1.Basic Laboratory Techniques 1.To cut a glass tube or glass rod, 2.To bend the glass rod at an angle, 3.To draw a glass jet from a glass tube 4.To bore a cork and fit a glass tube into it Viva-Voce 2.Characterisation and Purification of Chemical Substances 1.To determine the melting point of the given unknown organic compound and its identification (simple laboratory technique) Viva-Voce 2.To determine the boiling point of a given liquid when available in small quantity (simple laboratory method) Viva-Voce 3.To prepare crystals of pure potash alum [K<sub>2</sub>SO<sub>4</sub>.Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>.24H<sub>2</sub>O] from the given impure sample 4.To prepare the pure crystals of copper sulphate from the given crude sample 5.To prepare pure crystals of benzoic acid from a given impure sample Viva-Voce 3.Measurement of pH Values 1.To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper 2.To determine and compare the pH values of solutions of strong acid (HCl) and weak acid (CH<sub>3</sub>COOH) of same concentration 3.To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper 4.To study the pH change by common ion (CH<sub>3</sub>COO<sup>-</sup> ion) in case of weak acid (CH<sub>3</sub>COOH) 5.To determine the change in pH value of weak base (NH<sub>4</sub>OH) in presence of a common ion (NH<sub>4</sub><sup>+</sup>) Viva-Voce 4.Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2.To study the shift in equilibrium between [Co(H<sub>2</sub>O)<sub>6</sub>]<sup>2+</sup> and Cl<sup>-</sup> ions by changing the concentrations of either of the ions Viva-Voce 5. Quantitative Analysis 1.To prepare M/10 oxalic acid solution by direct weighing method 2.To prepare M/10 solution of sodium carbonate by direct weighing method 3.To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid 4.To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution Viva-Voce 6.Qualitative Analysis Analysis of Anions Analysis of Cations Viva-Voce 7.Detection of Elements in Organic Compounds 1.To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne’s test 2.To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number ..... by Lassaigne’s test Viva-Voce INVESTIGATORY PROJECTS 1.Checking of Bacterial Contamination in Water 1.To check the bacterial contamination in drinking water by testing sulphide ions Viva-Voce 2. Methods of Water Purification 1.To purify water from suspended impurities by using sedimentation 2. To purify water by boiling 3. o purify water by distillation method 4. To purify water by reverse osmosis technique 5. To purify water by GAC method 6. To purify water by bleach treatment 7. To purify water by oxidising agent 8. To purify water by ozone treatment method Viva-Voce 3. Water Analysis 1.To test the hardness of different water samples Viva-Voce 4. Foaming Capacity of Various Soaps 1 .To compare the foaming capacity of different washing soaps 2. To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva-Voce 5. Tea Analysis 1.To study the acidity of different samples of tea leaves (tea) by using pH paper Viva-Voce 6.Analysis of Fruits and Vegetable Juices 1. To analysis the fruit and vegetable juices for the constituent present in them Viva-Voce 7. Rate of Evaporation 1. To study the rate of evaporation of different liquids IViva-Voce 8. Effect of Acids and Bases on Tensile Strength of Fibres 1.To compare the tensile strength of natural fibres and synthetic fibres 2.To study the effect of acids and bases on tensile strength of different fibres Viva-Voce  
**Journal. Appendix** New Zealand. Parliament. House of Representatives 1898  
[Conceptual Chemistry Class XI Vol. I](#) S K Jain A book on Conceptual Chemistry  
**Annual Calendar of McGill College and University** McGill University 1890  
**Practical/Laboratory Manual Chemistry Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal** Dr. S. C. Rastogi 2020-06-23 An Excellent Book in Accordance with the latest syllabus for Class-11 Prescribed by CBSE/NCERT and Adopted by Various State Education Boards. (A) Basic Laboratory Techniques - 1. To cut a glass tube or glass rod, 2. To bend the glass rod at an angle, 3. To draw a glass jet from a glass tube, 4. To bore a cork and fit a glass tube into it. (B) Characterisation and Purification of Chemical Substances- 1. To determine the melting point of the given unknown organic compound and its identification (simple laboratory technique), 2. To determine the boiling point of a given liquid when available in small quantity (simple laboratory method), 3. To prepare crystals of pure potash alum [K<sub>2</sub>SO<sub>4</sub>.Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>.24H<sub>2</sub>O] from the given impure sample, 4. To prepare the pure crystals of copper sulphate from the given crude sample, 5. To prepare pure crystals of benzoic acid from a given impure sample. (C) Measurement of pH Values 1. To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper, 2. To determine and compare the pH values of solutions of strong acid (HCl) and weak acid (CH<sub>3</sub>COOH) of same concentration, 3. To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper, 4. To study the pH change by common ion (CH<sub>3</sub>COO<sup>-</sup> ion) in case of weak acid (CH<sub>3</sub>COOH), 5. To determine the change in pH value of weak base (NH<sub>4</sub>OH) in presence of a common ion (NH<sub>4</sub><sup>+</sup>), (D) Chemical Equilibrium 1. To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions, 2. To study the shift in equilibrium between [Co(H<sub>2</sub>O)<sub>6</sub>]<sup>2+</sup> and Cl<sup>-</sup> ions by changing the concentrations of either of the ions, (E) Quantitative Analysis 1. To prepare M/10 oxalic acid solution by direct weighing method, 2.To prepare M/10 solution of sodium carbonate by direct weighing method, 3.To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid, 4.To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution, (F) Qualitative Analysis 1. Analysis of Anions, 2. Analysis of Cations (G) Detection of Elements in Organic Compounds 1.To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne’s test, 2. To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number ..... by Lassaigne’s test INVESTIGATORY PROJECTS (A) Checking of Bacterial Contamination in Water 1.To check the bacterial contamination in drinking water by testing sulphide ions (B) Methods of Water Purification 1.To purify water from suspended impurities by using sedimentation, 2. To purify water by boiling, 3.To purify water by distillation method, 4.To purify water by reverse osmosis technique. 5.To purify water by GAC method, 6.To purify water by bleach treatment, 7.To purify water by oxidising agent, 8.To purify water by ozone treatment method. (C) Water Analysis 1. To test the hardness of different water samples. (D) Foaming Capacity of Various Soaps 1.To compare the foaming capacity of different washing soaps, 2.To study the effect of addition of sodium carbonate on foaming capacity of washing soap (E) Tea Analysis 1. To study the acidity of different samples of tea leaves (tea) by using pH paper (F) Analysis of Fruits and Vegetable Juices 1. To analyse the fruit and vegetable juices for the constituent present in them (G) Rate of Evaporation 1. To study the rate of evaporation of different liquids (H) Effect of Acids and Bases on Tensile Strength of Fibres 1.To compare the tensile strength of natural fibres and synthetic fibres, 2.To study the effect of acids and bases on tensile strength of different fibres. Log & Antilog Table  
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**ISC Chemistry Book 1 for Class XI (2021 Edition)** R.D.MADAN 2018 ISC Chemistry Book 1  
[Calendar - McGill University](#) McGill University 1904  
**Chemical news and Journal of physical science** 1773  
**Guy’s Hospital Gazette** 1889  
*Oswaal ISC Sample Question Paper Class 11 (Set of 5 Books) Physics, Chemistry, Biology, English 1 & 2 (For 2022 Exam)* Oswaal Editorial Board 2022-02-03 • 10 Sample Papers in each subject. 5 solved & 5 Self-Assessment Papers • All latest typologies Questions. • On-Tips Notes & Revision Notes for Quick Revision • Mind Maps for better learning  
[Calendar](#) University of Otago 1910

**Conceptual Chemistry Class XI Vol. II** S K Jain A book on Conceptual Chemistry  
*Chemistry* R P Manchanda A text book on Chemistry  
**ISC Practical Chemistry Class XI** S.P. Sharma & Dr Ajaya Baboo ISC Practical Chemistry Class XI  
[ISC Practical Chemistry Volume II for Class XII \(2021 Edition\)](#) Ajaya Baboo 2018 ISC Practical Chemistry Class XII has been thoroughly revised as per the latest syllabus for ISC (Class XII) prescribed by the Council for the Indian School Certificate Examinations (CICSE), New Delhi.  
**Chemistry Class 11 - [Bihar & JAC]** Dr. S. C. Rastogi, 2022-07-20 Syllabus : Unit I : Some Basic Concepts of Chemistry, Unit II : Structure of Atom, Unit III : Classification of Elements and Periodicity in Properties,Unit IV : Chemical Bonding and Molecular Structure, Unit V : States of Matter : Gases and Liquids, Unit VI : Chemical Thermodynamics, Unit VII : Equilibrium, Unit VIII : Redox Reactions, Unit IX : Hydrogen, Unit X : s-Block Elements (Alkali and Alkaline earth metals) Group 1 and Group 2 Elements, Unit XI : Some p-Block Elements General Introduction to p-Block Elements, Unit XII : Organic Chemistry—Some Basic Principles and Techniques, Unit XIII : Hydrocarbons Classification of Hydrocarbons, Unit XIV : Environmental Chemistry Content : 1. Some Basic Concepts of Chemistry, 2. Structure of Atom, 3. Classification of Elements and Periodicity in Properties, 4. Chemical Bonding and Molecular Structure, 5. States of Matter, 6. Thermodynamics, 7. Equilibrium, 8. Redox Reactions, 9. Hydrogen, 10. s-Block Elements 11. p-Block Elements, 12. Organic Chemistry—Some Basic Principles and Techniques 13. Hydrocarbons 14. Environmental Chemistry I. Appendix II. Log-antilog Table  
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*Report* New Zealand. Department of Education 1898  
*Oxford University Gazette* University of Oxford 1891  
[Oswaal CBSE Chapterwise & Topicwise Question Bank Class 11 Chemistry Book \(For 2022-23 Exam\)](#) Oswaal Editorial Board 2022-05-10 Chapter Navigation Tools • CBSE Syllabus : Strictly as per the latest CBSE Syllabus dated: April 21, 2022 Cir. No. Acad-48/2022 Latest Updates: Newly added topics/concepts has been included via dynamic code • Revision Notes: Chapter wise & Topic wise • Exam Questions: Includes Previous Years KVS exam questions • New Typology of Questions: MCQs, VSA,SA & LA including case based questions • NCERT Corner: Fully Solved Textbook Questions (Exemplar Questions in Physics, Chemistry, Biology) Exam Oriented Prep Tools • Commonly Made Errors & Answering Tips to avoid errors and score improvement • Mind Maps for quick learning • Concept Videos for blended learning • Academically Important (AI) look out for highly expected questions for the upcoming exams • Mnemonics for better memorisation • Self Assessment Papers Unit wise test for self preparation  
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[Comprehensive Practical Chemistry XI](#) Dr. N. K. Verma 2010-02  
**CBSE New Pattern Chemistry Class 11 for 2021-22 Exam (MCQs based book for Term 1)** Shahana Ansari 2021-09-10 1. This book deals with CBSE New Pattern Chemistry for Class 11 2. It is divided into 7 Chapters as per Term 1 Syllabus 3. Quick Revision Notes covering all the Topics of the chapter 4. Carries all types of Multiple Choice Questions (MCQs) 5. Detailed Explanation for all types of questions 6. 3 practice papers based on entire Term 1 Syllabus with OMR Sheet With the introduction of new exam pattern, CBSE has introduced 2 Term Examination Policy, where; Term 1 deals with MCQ based questions, while Term 2 Consists of Subjective Questions. Introducing, Arihant’s “CBSE New Pattern Series”, the first of its kind providing the complete emphasize on Multiple Choice Questions which are designated in TERM 1 of each subject from Class 9th to 12th. Serving as a new preparatory guide, here’s presenting the all new edition of “CBSE New Pattern Chemistry for Class 11 Term 1” that is designed to cover all the Term I chapters as per rationalized syllabus in a Complete & Comprehensive form. Focusing on the MCQs, this book divided the first have syllabus of Chemistry into 7 Chapters giving the complete coverage. Quick Revision Notes are covering all the Topics of the chapter. As per the prescribed pattern by the board, this book carries all types of Multiple Choice Questions (MCQs) including; Assertion - Reasoning Based MCQs and Cased MCQs for the overall preparation. Detailed Explanations of the selected questions help students to get the pattern and questions as well. Lastly, 3 Practice Questions are provided for the revision of the concepts. TOC Some Basic Concept of Chemistry, Structure of Atom, Classification of Elements and Periodicity in Properties, Chemical Bonding and Molecular Structure, Redox Reactions, Hydrogen, Organic Chemistry: Some Basic Principles and Techniques, Practice Papers (1-3).  
**The Chemical News and Journal of Industrial Science** 1862  
[Appendix to the Journals of the House of Representatives of New Zealand](#) New Zealand. Parliament. House of Representatives 1895  
*Competition Science Vision* 2003-08 Competition Science Vision (monthly magazine) is published by Pratiyogita Darpan Group in India and is one of the best Science monthly magazines available for medical

entrance examination students in India. Well-qualified professionals of Physics, Chemistry, Zoology and Botany make contributions to this magazine and craft it with focus on providing complete and to-the-point study material for aspiring candidates. The magazine covers General Knowledge, Science and Technology news, Interviews of toppers of examinations, study material of Physics, Chemistry, Zoology and Botany with model papers, reasoning test questions, facts, quiz contest, general awareness and mental ability test in every monthly issue.

*Comprehensive Practical Chemistry XII* Dr. N . K. Verma 2011-11-01

*Illustrated Guide to Home Chemistry Experiments* Robert Bruce Thompson 2012-02-17 For students, DIY hobbyists, and science buffs, who can no longer get real chemistry sets, this one-of-a-kind guide explains how to set up and use a home chemistry lab, with step-by-step instructions for conducting experiments in basic chemistry -- not just to make pretty colors and stinky smells, but to learn how to do real lab work: Purify alcohol by distillation Produce hydrogen and oxygen gas by electrolysis Smelt metallic copper from copper ore you make yourself Analyze the makeup of seawater, bone, and other common substances Synthesize oil of wintergreen from aspirin and rayon fiber from paper Perform forensics tests for fingerprints, blood, drugs, and poisons and much more From the 1930s through the 1970s, chemistry sets were among the most popular Christmas gifts, selling in the millions. But two decades ago, real chemistry sets began to disappear as manufacturers and retailers became concerned about liability. .em>The Illustrated Guide to Home Chemistry Experiments steps up to the plate with lessons on how to equip your home chemistry lab, master laboratory skills, and work safely in your lab. The bulk of this book consists of 17 hands-on chapters that include multiple laboratory sessions on the following topics: Separating Mixtures Solubility and Solutions Colligative Properties of Solutions Introduction to Chemical Reactions & Stoichiometry Reduction-Oxidation (Redox) Reactions Acid-Base Chemistry Chemical Kinetics Chemical Equilibrium and Le Chatelier's Principle Gas Chemistry Thermochemistry and Calorimetry Electrochemistry Photochemistry Colloids and Suspensions Qualitative Analysis Quantitative Analysis Synthesis of Useful Compounds Forensic Chemistry With plenty of full-color illustrations and photos, *Illustrated Guide to Home Chemistry Experiments* offers introductory level sessions suitable for a middle school or first-year high school chemistry laboratory course, and more advanced sessions suitable for students who intend to take the College Board Advanced Placement (AP)

Chemistry exam. A student who completes all of the laboratories in this book will have done the equivalent of two full years of high school chemistry lab work or a first-year college general chemistry laboratory course. This hands-on introduction to real chemistry -- using real equipment, real chemicals, and real quantitative experiments -- is ideal for the many thousands of young people and adults who want to experience the magic of chemistry.

*Arihant CBSE Chemistry Term 2 Class 11 for 2022 Exam (Cover Theory and MCQs)* Aditya Jangid 2021-11-20 With the newly introduced 2 Term Examination Pattern, CBSE has eased out the pressure of preparation of subjects and cope up with lengthy syllabus. Introducing Arihant's CBSE TERM II - 2022 Series, the first of its kind that gives complete emphasis on the rationalized syllabus of Class 9th to 12th. The all new "CBSE Term II 2022 - Chemistry" of Class 11th provides explanation and guidance to the syllabus required to study efficiently and succeed in the exams. The book provides topical coverage of all the chapters in a complete and comprehensive manner. Covering the 50% of syllabus as per Latest Term wise pattern 2021-22, this book consists of: 1. Complete Theory in each Chapter covering all topics 2. Case-Based, Short and Long Answer Type Question in each chapter 3. Coverage of NCERT, NCERT Exemplar & Board Exams' Questions 4. Complete and Detailed explanations for each question 5. 3 Practice papers based on the entire Term II Syllabus. Table of Content States of Matter: Gases and Liquids, Chemical Thermodynamics, Equilibrium, s - Block Element, Hydrocarbons, Practice Papers (1-3).

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**Chemist and Druggist** 1896

*Practical Chemistry* O. P. Pandey 1972